



Adsorption of Cobalamin onto Synthesized Carbon Nanotubes (CNT)

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ABSTRACT

The adsorption efficiency of synthesized carbon nanotubes at different doses and contact time were investigated. The amount of adsorbed dye increased with the increasing of CNTs dosage and reached a plateau region when the CNTs dosage achieved 0.175g L⁻¹. Equilibrium adsorption data were analyzed by Langmuir and Freundlich isotherm of the results revealed that Langmuir isotherm fitted the experimental results well. Kinetic analyzes were tested using pseudo-first order, pseudo second order and the intraparticle diffusion model. Kinetic studies showed that the adsorption kinetics were more accurately represented by a pseudo second order model.

Keywords: Adsorption, carbon nanotubes, titanium dioxide, cobalamin, kinetics, isotherms.
