



**Synthesis And Characterization of Spinel Ferrite**



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Accepted on 14<sup>th</sup> June 2014

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**ABSTRACT**

The spinel ferrite  $\text{Cu}_{1-x}\text{Fe}_x\text{Mn}_{1-x}\text{Co}_x\text{Cr}_{1-x}\text{Al}_x\text{O}_4$  where  $0 \leq x \leq 1$  has been prepared by the co-precipitation technique and is characterized by XRD, IR, Catalytic and saturation magnetization studies. All the compounds in the system form the single cubic spinel phase. IR spectra of the compounds show absorption bands in the region of  $500\text{-}1500\text{ cm}^{-1}$ . Catalytic studies using decomposition of  $\text{H}_2\text{O}_2$  as a model reaction between  $303\text{-}343\text{K}$  using first order rate law suggested lower catalytic power for the composition  $x = 0.00$  and then it increases gradually. The activation energy values calculated from catalytic studies between  $303\text{-}313\text{K}$  and  $333\text{-}343\text{K}$  are in the range of  $82.56\text{ KJ Mole}^{-1}$  to  $72.00\text{ KJ Mole}^{-1}$ . Saturation magnetization values calculated using  $2200$  gauss magnetic field, magnetization value ( $15\text{ emu gm}^{-1}$ ).

**Keywords:** Spinel ferrites, XRD, FTIR, Magnetic Hysteresis, Catalytic studies.

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