



## Determination of Eletriptan Hydro Bromide in pure and Pharmaceutical Formulations Using Cobalt Thiocyanide and Citric Anhydride by Spectrophotometric Method

M.L.N. Acharyulu<sup>1\*</sup>, P.V.S.R. Mohana Rao<sup>2</sup> and I. Siva Ramakoti<sup>3</sup>

1. Centurion University of Technology and Management, Andhra Pradesh- 530017, **INDIA**

2. Department of Engineering Chemistry, A.U.College of Engineering (A),  
Visakhapatnam-530003, **INDIA**

3. Department of Chemistry, Centurion University of Technology and Management, Odisha- 761211, **INDIA**  
Email: [acharyulu@cutmap.ac.in](mailto:acharyulu@cutmap.ac.in)

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### ABSTRACT

Two visible spectrophotometric methods were developed A and B for the determination of Eletriptan hydrobromide in pure and pharmaceutical formulations. Method A is based on the formation of coordination complex of tertiary amine of EHB (electron donor) and the central metal of the cobalt thiocyanate (acceptor) and Method B is based on internal salt formation involving aconitic anhydride (dehydration product of CiA) and the tertiary amine of EHB. The coloured products exhibit absorption  $\lambda_{max}$  at 623 nm and 546 nm for methods A and B respectively. Regression analysis of Beer-Lambert plots showed good correlation in the concentration ranges 4-24  $\mu\text{g L}^{-1}$ , correlation coefficients are 0.9886(A), 0.9877(B) respectively. The Sandell's sensitivities are  $2.7739 \times 10^{-3}$ ,  $1.9933 \times 10^{-3}$  ( $1 \text{ mole cm}^{-1}$ ) and molar absorptivity values are  $1.6706 \times 10^5$ ,  $2.3248 \times 10^5$  ( $\mu\text{g cm}^{-2}$ ). The proposed methods are applied to commercial available formulations and the results are statistically compared with those obtained by the UV reference method and validated by recovery studies.

### High Lights:

- The results are found satisfactory and reproducible.
- These methods are applied successfully for the estimation of the EHB in the presence of other ingredients that are usually present in formulations.
- These methods offer the advantages of rapidity, simplicity and sensitivity and low cost without the need for expensive instrumentation and reagents.

**Keywords:** Coordination complex, Dehydration Product, Tertiary Amino group, Regression Analysis.