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New Iridoid glycosides from Wendlandia puberula

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ABSTRACT

An extract of the leaves of Wendlandia puberula has yielded three new iridoid glycosides namely 8acetylharpagide, loganin and sweroside.

Keywords: 8-acetylharpagide, loganin and sweroside.

INTRODUCTION

Wendlandia puberula (rubiaceae) is known to yield iridoid glycosides[1-2]. The present communication deals with the isolation and characterization of three new iridoid glycosides designated as 8-acetylharpagide,1, loganin 2 and sweroside3 along with known compounds β -sitosterol and β -sitosterol- β -D-Glycoside. The structures of the compounds were determined from the studies of UV, IR and NMR spectra and chemical transformations.

MATERIALS AND METHODS

Extraction and isolation: The air-dried well chopped leaves of Wendlandia puberula (collected from Tharali, district Chamoli) was exhaustively defatted with light petroleum ether $(60-80^{\circ})$. The petroleum free mass was then extracted with 90% aqueous ethanol. The ethanol extract was concentrated to dry under reduced pressure in vacuum. A suspension of the residue was made with water, which was washed with diethyl ether for several times and then extracted with chloroform. The aqueous layer was successively extracted with EtOAc and BuOH (saturated with water). The EtOAc and BuOH layers were concentrated under reduced pressure to give EtOAc and BuOH extracts respectively. The BuOH extract was further treated with methanol: water (8:2) and filtered. The filtrate was evaporated to dry to give methanol extract. The light chloroform extract, ethyl acetate extract and BuOH were subjected to column chromatography over silica gel-G with various solvents in order to their increasing polarity. The CHCl₃ extract was subjected to CC over Sigel-G using gradient elution with C_6H_6 -EtOAc (100:0 \rightarrow 9:1) afforded β -sitosterol and β-Sitosterol-β-D-glycoside. The ethyl acetate extract was subjected to CC over Si-gel using gradient elution with CHCl₃:MeOH (1:0 \rightarrow 17:3) to get various fractions. The CHCl₃:MeOH (90:10) fraction was subjected to CC over Si-gel using gradient elution with CHCl₃:MeOH ($98:2\rightarrow90:10$) afforded compound 1 and various other fractions having mixture of compounds. The fractions obtained with CHCl₃:MeOH (93:7) were mixed and evaporated to dryness. The extract obtained was further chromatographed over

silica gel eluted with MeOH:CHCl₃ (9:1) afforded two fractions. Fraction 1 was subjected to CC over Sigel eluted with MeOH:CHCl₃ (9:1) afforded compound 1 and compound 2. Fraction 2 was subjected to repeated CC over silica gel eluted with MeOH:CHCl₃ (9:1) afforded compound 2 and compound 3.

Experimental: Melting points were determined on a MAC (MSW-403) apparatus and are uncorrected. Electronic spectra (EtOH) were obtained on a SP-3-200 Pye Unicam spectrophotometer as KBr pellets, NMR spectra were recorded in BRUKER WM 400 spectrometer (400 MHz for ¹H NMR and 100 MHz for ¹³C NMR). Chemical shifts are given on a ppm scale with TMS as an internal standard. The analytical data of the compounds were found satisfactory.

RESULTS AND DISCUSSION

The spectral results of three compounds are given below.

Compound 1: λ_{max} 224 and 308nm; ν_{max} 3500 (chelated OH), 2900 (C-H stretching of saturated carbon atom), 1700 and 1650 cm⁻¹ (α , β -unsaturated carbonyl function). ¹H-NMR (400 MHz, C₅D₅N): δ 6.71 (1H, d, J = 1.2 Hz, H-1), 6.53 (1H, d, J = 6.4 Hz, H-3), 5.15 (1H, dd, J = 6.4, 1.2 Hz, H-4), 4.03 (1H, d, J = 4.4 Hz, H-6), 2.06 (1H, dd, J = 14.8, 4.4 Hz, H-7 α), 2.54 (1H, d, J = 14.8 Hz, H-7 β), 3.54 (1H, brs, H-9), 1.59 (3H, s, H-10), 1.85 (3H, s, OAc), 5.30 (1H, d, J = 8 Hz, H-1'), 4.00 (1H, dd, J = 8.0, 8.8 Hz, H-2'), 3.99 (1H, t, J = 8.8 Hz, H-3'), 4.23 (1H, m, H-4'), 4.26 (1H, dd, J = 11.6, 2.4 Hz, H-5'), 4.50 (1H, dd, J = 11.6, 2.4 Hz, H-6'a), 4.32 (1H, dd, J = 11.6, 5.2 Hz, H-6'b). ¹³C-NMR (100 MHz, C₅D₅N): δ 94.7 (C-1), 142.2 (C-3), 108.1 (C-4), 73.2 (C-5), 78.7 (C-6), 45.8 (C-7), 87.2 (C-8), 55.1 (C-9), 22.6 (C-10), 99.1 (C-1'), 74.8 (C-2'), 78.4 (C-3'), 71.6 (C-4'), 76.8 (C-5'), 62.8 (C-6'), 22.1 (CH₃COO-) 170.9 (CH₃COO-).

Compound **2** : λ_{max} 224 and 308nm; ν_{max} 3450 (OH), 2900 (C-H stretching of saturated carbon atom), 1715 (unsaturated ester) and 1650 cm⁻¹ (enol ether).¹H-NMR (400 MHz, CD₃OD): δ 5.26 (1H, d, J = 4.4 Hz, H-1), 7.38 (1H, s, H-3), 3.09 (1H, m, H-5), 1.62 (1H, m, H-6a), 2.03 (1H, m, H-6b), 4.04 (1H, t, J = 4.4 Hz, H-7), 1.87 (1H, m, H-8), 2.23 (1H, m, H-9), 1.08 (3H, d, J = 6.8 Hz, H-10), 3.68 (3H, s, -OCH₃), 4.64 (1H, d, J = 8.0 Hz, H-1'), 3.17-3.39 (4H, m, H-2', H-3', H-4', H-5'), 3.66 (1H, dd, J = 11.8, 6.0 Hz, H-6'a), 3.89 (1H, dd, J = 11.8, 5.2 Hz, H-6'b). ¹³C-NMR (100 MHz, CD₃OD): δ 97.7 (C-1), 152.1 (C-3), 114.1 (C-4), 32.2 (C-5), 42.6 (C-6), 74.7 (C-7), 42.2 (C-8), 46.6 (C-9), 13.4 (C-10) 169.6 (-C-OO-), 51.7 (-COOCH₃), 99.9 (C-1'), 75.1 (C-2'), 78.3 (C-3'), 71.6 (C-4'), 78.0 (C-5'), 62.1 (C-6')

Compound **3** : λ_{max} 244 (iridoid enol ether system conjugated with a C-4 carbonyl group; ¹H-NMR (400 MHz, CD₃OD): δ 5.43 (1H, d, J = 1.2 Hz, H-I), 7.56 (1H, d, J = 2.4 Hz, H-3), 3.11 (1H, m, H-5), 1.71 (2H, m, H-6), 4.41 (2H, m, H-7), 5.56 (1H, m, H-8), 2.72 (1H, m, H-9), 5.33 (1H, dd, J = 2.0, 16.8 Hz, H-10a), 5.25 (1H, dd, J = 2.0, 9.0 Hz, H-10b), 4.63 (1H, d, J = 8.0 Hz, H-1'), 3.22 (1H, dd, J = 8.0, 8.8 Hz, H-2'), 3.71 (1H, t, J = 8.8 Hz, H-3'), 3.31 (1H, m, H-4'), 3.44 (1H, m, H-5'), 3.65 (1H, dd, J = 12.0, 6.0 Hz, H-6'a), 4.02 (1H, dd, J = 12.0, 6.4 Hz, H-6'b). ¹³C-NMR (100 MHz, CD₃OD): δ 98.1 (C-1), 154.3 (C-3), 105.3 (C-4), 27.3 (C-5), 25.1 (C-6), 70.1 (C-7), 132.1 (C-8), 42.6 (C-9), 121.2 (C-10), 169.7 (C-11), 99.4 (C-1'), 73.7 (C-2'), 76.2 (C-3'), 70.2 (C-4'), 72.1 (C-5'), 61.0 (C-6').

Compound 1: The molecular formula is $C_{17}H_{25}O_{11}$ and m.p. 191-193°C. The IR spectrum of displayed characteristic abosorption maxima at 3500 cm⁻¹ for a chelated OH group at 2900 cm⁻¹ for C-H stretching of saturated carbon atom, and at 1700 and 1650 cm⁻¹ for α,β -unsaturated carbonyl function. Its UV-spectrum showed absorption bands characteristic to an iridoid enol ether system[3] at 224 and 308 nm. The molecular formula of the compound **1** showed presence of five double bond equivalence in the molecule. The ¹H-NMR spectrum of coupled with detailed analysis of ¹H-¹H COSY indicated presence of two integrated protons signals each for 1H at δ 6.53 (d, J = 6.4 Hz, H-3), and 5.15 (dd, J = 6.4, 1.2 Hz, H-4), two oxygen bearing methine signal at δ 6.71 (d, J = 1.2 Hz, H-1), methylene protons at δ 2.06 (1H, dd, J

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= 14.8, 4.4 Hz, H-7 α), and 2.54 (1H, d, J = 14.8 Hz, H-7 β), one oxygen bearing methine proton signal at δ 4.03 (1H, d, J = 4.4 Hz, H-6), and methine proton signal at 3.54 (1H, brs, H-9). In addition to this a three protons singlet at δ 1.85 assignable for acetoxy group, methyl protons singlet at δ 1.59, and seven protons due to the sugar moiety were observed. A methylene proton signal at δ 2.06 (dd) showed coupling with another methylene proton at δ 2.54 and with an oxygen bearing methine proton at δ 4.03. A methine proton signal (brs) showed weak coupling with the proton signal appeared at δ 6.71 which might be due to a methine proton bearing two oxygen atoms and is compatible with the H-1 proton signal of most of the iridoids having O-glycosylation at C-1 carbon [3-6]. The ¹³C-NMR spectrum and distortion less enhancement by polarization transfer (DEPT) spectrum showed 17 carbon signals; three quaternary carbon, 10 methine carbons, two methylene carbon and two methyl carbons. Acid hydrolysis of compound 1 with 5% HCl gave a sugar which was identified as D-glucose by paper chromatography. The glycosidic nature of the compound was supported by a doublet at δ 5.30 (J = 8.0 Hz) assignable to the anomeric proton of β -D-glucose. The usual location of sugar moiety at position O-1 of the aglycone was shown by the downfield shifted signal[4] of H-1 (δ 6.71, d, J = 1.2 Hz). The ¹³C-NMR chemical shifts of anomeric carbon atom (C-1') at δ 99.1 and the chemical shifts of other carbon atoms of sugar moiety [δ 74.8 (C-2'), 78.4 (C-3'), 71.6 (C-4'), 76.8 (C-5'), 62.8 (C-6')] are in agreement with the NMR spectrum and thus confirmed the presence of glucose in the molecule. The ¹³C-NMR spectrum confirmed the presence of methyl function (δ 22.6), a methylene carbon [δ 45.8 (C-7)], two methine carbons [(δ 78.7 (C-6) having oxygen function and 55.1 (C-9)], a secondary carbinyl carbon [δ 94.7 (C-1)], two quaternary carbons having an oxygen function [δ 73.2 (C-5), and 87.2 (C-8)], a di-substituted double bond [δ 142.2 (C-3), 108.1 (C-4),] and an acetate function $[(\delta 22.1 \text{ (CH}_3\text{COO-)}) 170.9 \text{ (-COO-)}]$. These spectral data was strongly reminiscent of that reported for 8-acetylharpagide [4]. On the basis of above discussed spectral data compound 1 was identified as 8-acetylharpagide (structure 1).



Structure of Compound 1

Compound 2: The molecular formula is $C_{17}H_{26}O_{10}$, m.p. 220-222⁰C. The exhibited IR absorption characteristic for hydroxyl group (at 3450 cm⁻¹), C-H stretching of saturated carbon atom (2900 cm⁻¹), unsaturated ester (1715 cm⁻¹) and enol ether at 1650 cm⁻¹. UV-spectrum showed characteristic absorption bands at 224 and 308 nm for an iridoid enol ether system conjugated with a C-4 carboxyl group[3]. The ¹H-NMR spectrum of compound 2 indicated presence of 26 proton signals. An isolated singlet (1H) at δ 7.38 was assigned for a proton attached at tri-substituted double bond. An oxygen bearing methine proton signal at δ 4.04 for 1H appeared as a triplet (J = 4.4 Hz, H-7), and three methine signals at δ 3.09 (1H, m, H-5), 1.87 (1H, m, H-8), and 2.23 (1H, m, H-9), was assignable for H-5, H-8 and H-9 protons of iridoids respectively. Two methylene proton at δ 1.62 and 2.03 each for 1H appeared as a multiplet was assigned for H-6a and H-6b. In addition to this a three protons singlet at δ 3.68 assinable for protons of a carboxyl group, a doublet for three protons of a methyl function at δ 1.08, and seven protons due to the sugar moiety were observed. A doublet (J = 4.4 Hz) appeared at δ 5.26 was corroborated with the H-1 proton signal of most of the iridoids having O-glycosylation at C-1 carbon[4-6]. The ¹³C-NMR spectrum and distortion less enhancement by polarization transfer (DEPT) spectrum showed 17 carbon signals; two quaternary carbons. 11 methine carbons, two methylene carbon and two methyl carbons. Acid hydrolysis of compound 2 with 5% HCl gave a sugar, which was identified as D-glucose by paper chromatography. The glycosidic nature of the compound was supported by a doublet at δ 4.64 (J = 8.0 Hz) assignable to the anomeric proton of β -D-glucose. The ¹³C-NMR chemical shifts of anomeric carbon atom (C-1') at δ 99.9 and the chemical shifts of other carbon atoms of sugar moiety [8 75.1 (C-2'), 78.3 (C-3'), 71.6 (C-4'), 78.0 (C-5'), 62.1 (C-6')] are in agreement with the ¹H-NMR spectrum and thus confirmed the presence of glucose in the molecule. The usual location of sugar moiety at position O-1 of the aglycone was shown by the downfield shifted signal of H-1 (δ 5.26, d, J = 4.4 Hz). The ¹³C-NMR spectrum confirmed the presence of methyl function (δ 13.6), a methylene carbon [6 42.6 (C-6)], three methine carbons [(6 32.2 (C-5), 42.2 (C-8), and 46.6 (C-9)], a methine group having oxygen function at δ 74.7, a secondary carbinyl carbon [δ 97.7 (C-1)], a trisubstituted double bond [δ 152.1 (C-3), 114.1 (C-4),] and an acetate function [(δ 13.6 (-COOCH₃) 169.6 (-COO-)]. The location of methyl group was determined at position C-8 by the 1H-NMR spectrum in which a methine proton appeared as a multiplet due to coupling with three methyl protons, and two methine protons (H-7 and H-9). The downfield chemical shifts of a methine proton at δ 4.04 attributed for H-7 indicated that hydroxyl function was attached at C-7 position. The UV absorption coupled with the ¹Hand ¹³C-NMR data establishes that a carboxyl group is located at C-4 carbon. On the basis of above discussed spectral data compound 2 was identified as loganin (Structure-2) which was further confirmed comparison of spectral data with that of reported data [7-9].



Structure of compound 2

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Compound 3 : The molecular formula is $C_{16}H_{22}O_9$ showed a typical iridoid colour reaction with hydrochloric acid. UV-spectrum of compound 3 showed characteristic absorption bands at 244 nm for an iridoid enol ether system conjugated with a C-4 carbonyl group [3]. The detailed analysis of ¹H-NMR spectrum coupled with of ${}^{1}\text{H}{}^{-1}\text{H}$ COSY indicated presence of two methylene protons at δ 1.71 (multiplet for 2 protons, H-6) which showed coupling with a methylene proton signals (multiplet for 2 protons) resonated downfield (δ 4.41, H-7) in comparison with H-6 indicated that the later is attached with an oxygen function. The methylene proton signal at δ 1.71 also showed coupling with a methine proton signal at δ 3.11 (*m*, 1H, H-5) which in turn showed coupling with a methine proton signal at δ 5.43 (*d*, *J* = 1.2 Hz, H-1) and at δ 2.72 (1H, m, H-9). Detailed analysis of ¹H-¹H-COSY showed that the methine proton signal attached to a mono-substituted double bond resonated at δ 5.56 (1H, m, H-8) showed coupling with a methine proton appeared at δ 2.72 (H-9) and with two methylene protons appeared at δ 5.33 (1H, dd, J = 2.0, 16.8 Hz, H-10a), and δ 5.25 (1H, dd, J = 2.0, 9.0 Hz, H-10b). These data indicated presence of a vinyl group in the molecule. An integrated methine proton signal which appeared as a double (J = 2.4) at δ 7.56 showed long range coupling in ¹H-¹H COSY spectrum with the methine proton signal appeared as a multiplet at δ 3.11. A doublet (J = 1.2 Hz) appeared at δ 5.43 was corroborated with the H-1 proton signal of most of the iridoids having O-glycosylation at C-1 carbon [4-6]. Besides these protons signals the ¹H-NMR showed a doublet at δ 4.63 (1H, d, J = 8.0 Hz, H-1'), which was corroborated with the presence of a β -D-glucose moiety in the molecule. The above discussed ¹H-NMR data are strongly reminiscent with those reported for sweroside [7-9]. The ¹³C-NMR spectrum of compound 3 and DEPT spectrum showed 16 carbon signals out of which 2 are quaternary, 10 methine, and four methylene carbons. The assignment of methylene, methine and quaternary carbon signals readily made by HMQC experiment. On acid hydrolysis with acid, yielded an aglycon and D-glucose. The glycosidic nature of the compound was supported by a doublet at δ 4.63 (J = 8.0 Hz) assignable to the anomeric proton of β -D-glucose. The ¹³C-NMR chemical shifts of anomeric carbon atom (C-1') at δ 99.4 and the chemical shifts of other carbon atoms of sugar moiety are in agreement with the ¹H-NMR spectrum and thus confirmed the presence of glucose in the molecule. The usual location of sugar moiety at position O-1 of the aglycone was shown by the downfield shifted signal of H-1 (δ 5.43). The ¹³C-NMR data of compound 3 are in agreement with the ¹H-NMR data. The presence of a vinyl function as deduced by ¹H-NMR was confirmed by ¹³C chemical shifts of unsaturated carbon atom at δ 132.1 (C-8) and δ 121.2 (C-10). The ¹³C-NMR spectrum also showed presence of two methine carbons at δ 27.3 (C-5) and 42.6 (C-9), a methylene carbon at δ 25.1 (C-6), a methylene carbon bearing an oxygen at δ 70.1 (C-7), a secondary carbinyl carbon [δ 98.1 (C-1)], a trisubstituted double bond [δ 154.3 (C-3), 105.3 (C-4),] and a carbonyl function at δ 169.7. The location of vinyl group was determined at position C-9 by the ¹H-NMR spectrum in which a methine proton appeared as a multiplet at δ 2.72 (H-9) showed coupling with three methine protons appeared at δ 5.43 (1H, d, J = 1.2 Hz, H-1), 3.11 (1H, m, H-5), and 5.56 (1H, m, H-8). The UV absorption coupled with the 1 H- and 13 C-NMR data establishes that a carbonyl group is located at C-4 carbon. On the basis of above discussed spectral data compound 3 was characterized as sweroside (Structure-3) which was confirmed by comparison of spectral data with the reported data [7-9].



Structure of Compound 3

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APPLICATIONS

The extracted compounds loganin, sweroside are used in traditional Chinese medicine. They are also used for the treatment of such ailments as acute fever, headache, respiratory infection, and epidermic diseases. 8-O-acetylharpagide has antioxidant Activity and Vasoconstrictor activity.

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