



## Synthesis, Characterisation And Antimicrobial Activities of Schiff Bases containing 1,3,4-oxadiazole Moiety

Jaydeep A. Patel and Navin B. Patel\*

\*Department of Chemistry, Veer Narmad South Gujarat University, Surat-395007, **INDIA**

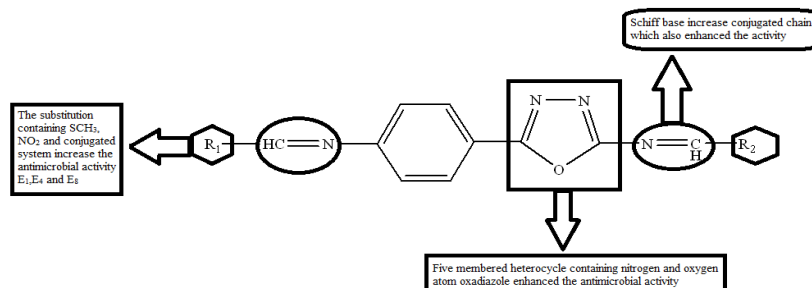
Email: [drnavinbpatel@gmail.com](mailto:drnavinbpatel@gmail.com), [jaydeep073@gmail.com](mailto:jaydeep073@gmail.com)

Accepted on 19<sup>th</sup> December 2017, Published online on 27<sup>th</sup> January 2018

### ABSTRACT

In present work PABA A reacted with substituted aldehyde B to give Schiff base C. Which further reacted with semicarbazide and substituted aldehyde D to obtain cyclic product and Schiff base as final moiety N-(4-(5-(substitutedamino)-1,3,4-oxadiazole-2-yl) phenyl) substitutedimine E. The synthesized molecules were spectrally evaluated by IR, <sup>1</sup>H NMR, <sup>13</sup>C NMR and Mass spectra. Synthetic analogues screened for their antimicrobial activities. Among the tested compounds, E<sub>8</sub> displayed very good antibacterial activity against E. coli and S. aureus strain. E<sub>4</sub>, E<sub>5</sub> and E<sub>8</sub> showed very good antibacterial activity against P. aeruginosa strain. E<sub>1</sub> has given good antifungal activity against C. albicans species. E<sub>7</sub> and E<sub>8</sub> observed good antifungal activity against A. niger and A. clavatus species.

### Graphical Abstract:



Synthesis of N-(4-(5-(substitutedamino)-1,3,4-oxadiazole-2-yl)phenyl) substitutedimine.

**Keywords:** Oxadiazole, schiff base, antibacterial and antifungal.

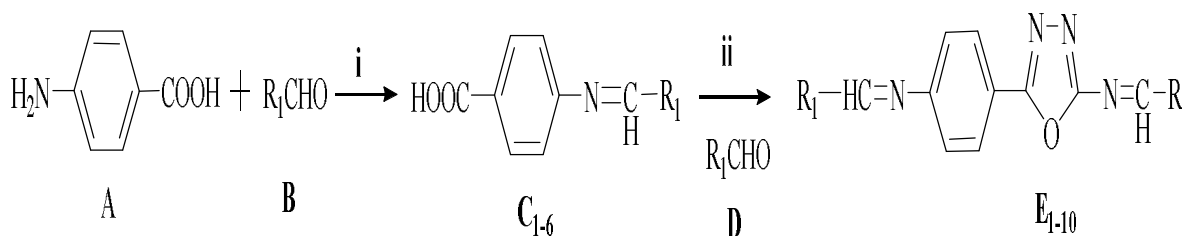
## INTRODUCTION

Oxadiazole, a heterocyclic molecule containing one oxygen and two nitrogen atoms in a five-membered ring [1]. Which has attracted a wide attention for Researcher as a new therapeutic agent. Several methods have been revealed in the literature for the synthesis of 1,3,4-oxadiazoles and out of various isomers particularly 1,3,4-oxadiazole derivatives exhibit wide range of biological activities. Also various route for

the synthesis of 1,3,4-oxadiazole have been reported[2-7]. Acid hydrazides [8] have been in general use as the starting materials in some 1,3,4-oxadiazole.

Researchers have already reported that gram positive bacteria are much more susceptible to antimicrobial agents as compared to gram negative bacteria. These differences may be attributed to the fact that the cell wall in gram positive bacteria is of single layer whereas the gram-negative bacteria have multilayered cell wall. 1,3,4-oxadiazoles have significant pharmaceutical interest. It associated considerable biological activities such as antifungal [9], antibacterial [10], analgesic [11-13], antitubercular [14], cytotoxicity [15] and antioxidant agent [16,17].

Schiff bases, known as Imines are compounds containing azomethine group  $-(HC=N)-$  which imports in elucidating the mechanism of transformation and racemization reaction in biological system [18] and represented by the general formula  $R_3R_2C=NR_1$ . They are the condensed products of aldehydes or ketones and were first reported by Hugo Schiff in 1864 [19]. Originally, the classical synthetic route for synthesis of Schiff bases was reported by Schiff which involves condensation of primary amines with carbonyl compounds [20] under azeotropic distillation with the removal of water. Interests in these compounds are largely due to their relatively simple procedures of synthesis as well as significant biological activity. They are well known intermediate for the preparation of azetidinone, thiazolidinone, arylacetamide, metal complexes and many other derivatives [21]. Schiff's bases have been playing vital roles in pharmaceuticals, rubber additives, as amino protective groups in the synthetic organic chemistry and several biologically active organic compounds. They are also used as liquid crystals in analytical, medicinal, and polymer chemistry [22]. The Schiff bases possessing diversified biological applications such as antitubercular, anticancer, antibacterial, antifungal, antitumor, anthelmintic, anti-HIV, antiparasitic activities [23, 24].



**Scheme 1.** Synthesis of N-(4-(5-(substitutedeneamino)-1,3,4-oxadiazole-2-yl) phenyl) substitutedimine. Experimental conditions: (i)Methanol, 2-3 drops acetic acid, 2-4 h heating with stirring; (ii) Methanol, Semicarbazide, 6-8 h reflux;  $4R_1 = -C_8H_7, -C_6H_4-4-C_3H_7, -C_4H_3S, -C_6H_4-SCH_3, -C_6H_4-4-NO_2, -C_4H_3O$ ;  $4R_2 = -C_5H_4N, -C_4H_3O, -C_6H_4-SCH_3, -C_6H_4-4-C_3H_7, -C_8H_7$ .

## MATERIALS AND METHODS

All the chemicals were used as of analytical grade. Melting points were determined on a Fisher–Johns Melting Point apparatus. A purity of the compounds was checked by TLC on silica gel plates with visualization by UV-light and iodine chamber.  $^1H$ -NMR and  $^{13}C$ -NMR spectra were recorded on Bruker Avance II 400 MHz spectrometer instruments in DMSO- $d_6$  and chemical shifts were recorded in parts per million downfield from tetramethylsilane. Mass spectra were recorded on LC-MS Agilent 1100 series.

**General synthesis of 4-(substitutedeneamino) benzoic acid ( $C_{1-6}$ ):** Equimolar solution of 4-amino benzoic acid (A) (0.1mol) and cinnamaldehyde aldehyde ( $B_1$ ) (0.1mol) in ethanol taken in a Borosil conical flask. Then add few drops of acetic acid as catalyst. Put the reaction on electric starrer for stirring with heating till precipitate obtained [25]. The progress of the reaction was monitored by TLC using n-hexane: ethyl acetate (7:1), after every 30 min. On completion of the reaction, the reaction mixture was

cooled at room temperature and was poured into ice. The product was washed by distil water. It was dried and purified by recrystallization from ethanol to obtained analytical sample (C<sub>1</sub>). Other compounds (C<sub>2-6</sub>) were prepared by same method.

**Table 1.** Physical data of compounds (C<sub>1-6</sub>)

Sr. No.	Code	Yield□	M.P. <sup>0</sup> C	M.F.
1.	C <sub>1</sub>	84	153	C <sub>16</sub> H <sub>13</sub> O <sub>2</sub> N
2.	C <sub>2</sub>	82	146	C <sub>17</sub> H <sub>17</sub> O <sub>2</sub> N
3.	C <sub>3</sub>	87	130	C <sub>12</sub> H <sub>9</sub> O <sub>2</sub> NS
4.	C <sub>4</sub>	78	142	C <sub>15</sub> H <sub>15</sub> O <sub>2</sub> NS
5.	C <sub>5</sub>	85	159	C <sub>14</sub> H <sub>12</sub> O <sub>4</sub> N <sub>2</sub>
6.	C <sub>6</sub>	83	134	C <sub>12</sub> H <sub>9</sub> O <sub>3</sub> N

**General Synthesis of N-(4-(5-(substitutedeneamino)-1,3,4-oxadiazole-2-yl) phenyl) substitutedimine (E<sub>1-10</sub>):** To a mixture of compound (C<sub>1</sub>) (0.01mol), semicarbazide (0.012mol) and Pyridine 2-carboxaldehyde (D<sub>1</sub>) in ethanol in a Borosil conical flask was put on starrer for starring with heating at 90<sup>0</sup>C for 5-6 h [26]. The TLC was checked after 1.5 h. After completion of reaction cooled the reaction mixture at room temperature and then pour over ice cold water. The precipitate obtained was wash with water followed by distilled water. Then precipitated was crystallised by ethanol to make pure product (E<sub>1</sub>) as analytical sample. Other compounds (E<sub>2-6</sub>) were prepared by same method.

**Table 2.** Physical data of compounds (E<sub>1-10</sub>)

Sr. No.	Code	Yield□	M.P. <sup>0</sup> C	M.F.
1.	E <sub>1</sub>	62	254	C <sub>23</sub> H <sub>17</sub> ON <sub>5</sub>
2.	E <sub>2</sub>	68	237	C <sub>23</sub> H <sub>20</sub> O <sub>2</sub> N <sub>4</sub>
3.	E <sub>3</sub>	54	224	C <sub>21</sub> H <sub>16</sub> ON <sub>4</sub> S <sub>2</sub>
4.	E <sub>4</sub>	56	220	C <sub>26</sub> H <sub>24</sub> ON <sub>4</sub>
5.	E <sub>5</sub>	51	194	C <sub>25</sub> H <sub>21</sub> O <sub>3</sub> N <sub>5</sub>
6.	E <sub>6</sub>	64	247	C <sub>19</sub> H <sub>13</sub> ON <sub>5</sub>
7.	E <sub>7</sub>	58	217	C <sub>21</sub> H <sub>15</sub> O <sub>2</sub> N <sub>4</sub>
8.	E <sub>8</sub>	69	204	C <sub>25</sub> H <sub>20</sub> ON <sub>4</sub>
9.	E <sub>9</sub>	73	242	C <sub>20</sub> H <sub>13</sub> O <sub>4</sub> N <sub>5</sub>
10.	E <sub>10</sub>	57	191	C <sub>24</sub> H <sub>17</sub> O <sub>3</sub> N <sub>5</sub>

## RESULTS AND DISCUSSION

Synthetic steps given in a scheme in which all reaction steps completion was checked by TLC using sufficient solvents. Products obtained at different steps were purified by crystallisation using particular solvents. All intermediates were characterised by IR and <sup>1</sup>H NMR. Final products were identified by IR, <sup>1</sup>H NMR, <sup>13</sup>C NMR and Mass spectral studies. All final moieties were screened for antimicrobial activity.

### Spectral characterization of compounds

**C<sub>5</sub>:** IR (KBr)  $\nu$  cm<sup>-1</sup>: 3058 (OH), 1710 (C=O), 1616 (C=N); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$  (ppm): 10.47 (s, 1H, OH), 8.73 (s, 1H, =CH), 7.65-7.15 (m, 8H, Ar-H).

**C<sub>6</sub>:** IR (KBr)  $\nu$  cm<sup>-1</sup>: 3062 (OH), 1696 (C=O), 1604 (C=N); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$  (ppm): 10.15 (s, 1H, OH), 8.56 (s, 1H, =CH), 7.75-7.17 (m, 8H, Ar-H).

**E<sub>1</sub>:** IR (KBr)  $\nu$  cm<sup>-1</sup>: 2976, 2887 (CH), 1670 (C=C), 1645 (N-N), 1598 (C=N), 1092 (C-O-C); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$  (ppm): 8.75 (d, 1H, Ar-H pyridine), 8.53 (s, 1H, =CH), 8.06 (s, 1H, =CH), 7.87-7.66, (m, 3H, Ar-H pyridine), 7.52-7.26, (m, 9H, Ar-H), 7.18 (d, 1H, =CH), 6.65 (t, 1H, =CH). <sup>13</sup>C NMR(100MHz, DMSO-d<sub>6</sub>)  $\delta$  (ppm): 163.83, 162.86 (=CH), 158.65 (C<sub>5</sub>, Pyridine), 156.80 (C=N, Oxadiazole), 153.14 (C<sub>2</sub>, Pyridine), 149.44 (aromatic ring), 146.71 (C<sub>3</sub>, pyridine), 142.96 (C<sub>1</sub>, pyridine), 139.62 (C=N, Oxadiazole),

136.25 (C=C), 128.85, 126.29, 122.69, 118.22, 114.35, 111.05, (aromatic ring), 104.52 (C=C); ESI-MS: m/z calculated 379.42, found [M+H]<sup>+</sup> 379.19.

**E<sub>2</sub>**: IR (KBr)  $\nu$  cm<sup>-1</sup>: 3157, 3034 (CH<sub>3</sub>), 2863 (CH<sub>2</sub>), 1648 (N-N), 1595 (C=N), 1093 (C-O-C); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$  (ppm): 8.53 (s, 1H, =CH), 8.06 (s, 1H, =CH), 7.83-7.18 (m, 8H, Ar-H), 6.87-6.53 (m, 3H, Ar-H furfural), 2.87 (t, 2H, CH<sub>2</sub>), 1.64 (m, 2H, CH<sub>2</sub>), 1.18 (d, 3H, CH<sub>3</sub>); <sup>13</sup>C NMR(100MHz, DMSO-d<sub>6</sub>)  $\delta$  (ppm): 158.65 (=CH), 156.80 (C=N, Oxadiazole), 153.13 (=CH), 149.44 (aromatic ring), 146.71 (C<sub>4</sub>, furfural), 142.96 (aromatic ring), 139.62 (C=N, Oxadiazole), 136.25, 128.85, 122.69(C<sub>2</sub>, furfural), 118.23(C<sub>3</sub>, furfural), 114.35, 111.05, 100.46 (aromatic ring), 40.95 (CH<sub>2</sub>), 25.54 (CH<sub>2</sub>), 13.49 (CH<sub>3</sub>); ESI-MS: m/z calculated 384.43, found [M+H]<sup>+</sup> 384.56.

**E<sub>3</sub>**: IR (KBr)  $\nu$  cm<sup>-1</sup>: 3065 (CH<sub>3</sub>), 1646 (N-N), 1602 (C=N), 1139 (C-S), 1092 (C-O-C), 814 (C-S-C); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$  (ppm): 8.80 (s, 1H, =CH), 8.36 (s, 1H, =CH), 7.71-7.21 (m, 8H, Ar-H), 7.17-6.84 (m, 3H, Ar-H thiophene), 2.38 (s, 3H, CH<sub>3</sub>). <sup>13</sup>C NMR(100MHz, DMSO-d<sub>6</sub>)  $\delta$  (ppm): 160.00 (=CH), 157.23 (C=N, Oxadiazole), 152.86 (=CH), 148.24 (aromatic ring), 144.34 (C<sub>1</sub>, thiophene), 142.96 (aromatic ring), 138.56(C=N, Oxadiazole), 132.25, 130.14(C<sub>2</sub>, furfural), 129.41, 128.85, 128.50(C<sub>4</sub>, furfural), 127.23(C<sub>3</sub>, furfural), 126.92, 126.29, 124.82, 122.37, 114.35, 111.05, (aromatic ring), 14.87 (SCH<sub>3</sub>); ESI-MS: m/z calculated 372.45, found [M+H]<sup>+</sup> 372.78.

**E<sub>4</sub>**: IR (KBr)  $\nu$  cm<sup>-1</sup>: 3183, 3068 (CH<sub>3</sub>), 2870 (CH<sub>2</sub>), 1646 (N-N), 1603 (C=N), 1139 (C-S), 1093 (C-O-C); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$  (ppm): 8.63 (s, 1H, =CH), 8.14 (s, 1H, =CH), 7.62-7.18 (m, 12H, Ar-H), 2.78 (t, 2H, CH<sub>2</sub>), 2.27 (s, 3H, CH<sub>3</sub>), 1.61 (m, 2H, CH<sub>2</sub>), 1.14 (d, 3H, CH<sub>3</sub>); <sup>13</sup>C NMR(100MHz, DMSO-d<sub>6</sub>)  $\delta$  (ppm): 158.46 (=CH), 156.67 (C=N, Oxadiazole), 152.93 (=CH), 149.21 (aromatic ring), 143.26, 142.07 (aromatic ring), 139.51 (C=N, Oxadiazole), 136.25, 132.8, 129.4, 128.85, 126.9 114.35, 111.05, (aromatic ring), 41.15 (CH<sub>2</sub>), 25.72 (CH<sub>2</sub>), 14.64 (SCH<sub>3</sub>), 13.49 (CH<sub>3</sub>); ESI-MS: m/z calculated 408.49, found [M+H]<sup>+</sup> 408.06.

**E<sub>5</sub>**: IR (KBr)  $\nu$  cm<sup>-1</sup>: 3173, 3053 (CH<sub>3</sub>), 2856 (CH<sub>2</sub>), 1644 (N-N), 1608 (C=N), 1535, 1353 (NO<sub>2</sub>), 1139 (C-S), 1091 (C-O-C); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$  (ppm): 8.59 (s, 1H, =CH), 8.10 (s, 1H, =CH), 7.68-7.18 (m, 12H, Ar-H), 2.75 (t, 2H, CH<sub>2</sub>), 1.66 (m, 2H, CH<sub>2</sub>), 1.14 (d, 3H, CH<sub>3</sub>); <sup>13</sup>C NMR(100MHz, DMSO-d<sub>6</sub>)  $\delta$  (ppm): 158.52 (=CH), 156.59 (C=N, Oxadiazole), 152.93 (=CH), 149.25 (aromatic ring), 142.86, 141.74 (aromatic ring), 139.51 (C=N, Oxadiazole), 136.25, 132.78, 129.36, 128.85, 126.9 114.35, 111.15, (aromatic ring), 41.26 (CH<sub>2</sub>), 25.64 (CH<sub>2</sub>), 13.49 (CH<sub>3</sub>); ESI-MS: m/z calculated 439.49, found [M+H]<sup>+</sup> 439.78.

## APPLICATIONS

The disc diffusion method was used for the screening of antimicrobial activity. Antibacterial activity of N-(4-(5-(substitutedeneamino)-1,3,4-oxadiazole-2-yl) phenyl) substitutedimine (E<sub>1-10</sub>) was screened against two species *E. coli* and *P. Aeruginosa* Gram-positive bacteria *i.e.* two species of *S. aureus* and *S. pyogenes* as Gram-negative bacteria. The antifungal activity of the compounds was tested against two fungal organisms namely *C. albicans*, *A. niger* and *A. clavatus*. Antibacterial studies were compared against chloramphenicol, ciprofloxacin, norfloxacin and ampicillin as standard drugs. Nystatin and Griseofulvin were used for antifungal activity studies as reference compounds. From screened Compounds E<sub>8</sub> was found to be excellent active and E<sub>1</sub> good active against *E. coli*. Molecules E<sub>4</sub>, E<sub>5</sub>, and E<sub>8</sub> exhibited excellent activity and E<sub>7</sub> good activity against *P. aeruginosa*. Compounds showed excellent potency was E<sub>8</sub> and E<sub>2</sub>. E<sub>4</sub>, E<sub>5</sub>, E<sub>7</sub> were good potency against *S. aureus*. Moieties E<sub>1</sub>, E<sub>4</sub>, E<sub>5</sub>, E<sub>6</sub>, E<sub>7</sub> and E<sub>8</sub> exhibited considerable activity against *S. pyogenes*, when compared with Standard drugs. Derivatives found to be potent were E<sub>1</sub>, E<sub>4</sub>, E<sub>5</sub>, E<sub>9</sub> and E<sub>10</sub> against *C. albicans*. Compound E<sub>7</sub> and E<sub>8</sub> possessed moderate activity against *A. niger* and molecule E<sub>7</sub> and E<sub>8</sub> demonstrated moderate activity against *A. clavatus*, when compared with reference drugs.

**Table 3.** Antibacterial activity compounds (E<sub>1-10</sub>)

MINIMAL BACTERICIDAL CONCENTRATION( $\mu\text{gmL}^{-1}$ )				
Compound No.	<i>E. COLI</i>	<i>P. AERUGINOSA</i>	<i>S. AUREUS</i>	<i>S. PYOGENUS</i>
	MTCC 443	MTCC 1688	MTCC 96	MTCC 442
E <sub>1</sub>	62.5	500	125	100
E <sub>2</sub>	125	125	100	125
E <sub>3</sub>	250	250	250	200
E <sub>4</sub>	250	50	100	100
E <sub>5</sub>	125	25	62.5	100
E <sub>6</sub>	100	125	125	100
E <sub>7</sub>	125	100	100	62.5
E <sub>8</sub>	25	25	50	100
E <sub>9</sub>	250	250	500	500
E <sub>10</sub>	250	250	500	125
Drug	Micromolar ( $\mu\text{gmL}^{-1}$ )			
AMPICILLIN	100	-	250	100
CHLORAMPHENICOL	50	50	50	50
CIPROFLOXACIN	25	25	50	50
NORFLOXACIN	10	10	10	10

**Table 4.** Antifungal activity compounds (E<sub>1-10</sub>)

MINIMAL FUNGICIDAL CONCENTRATION( $\mu\text{gmL}^{-1}$ )			
Compound No.	<i>C. ALBICANS</i>	<i>A. NIGER</i>	<i>A. CLAVATUS</i>
	MTCC 227	MTCC228	MTCC1323
E <sub>1</sub>	250	>1000	>1000
E <sub>2</sub>	1000	>1000	>1000
E <sub>3</sub>	1000	>1000	>1000
E <sub>4</sub>	500	>1000	>1000
E <sub>5</sub>	500	500	1000
E <sub>6</sub>	1000	1000	1000
E <sub>7</sub>	1000	250	250
E <sub>8</sub>	1000	250	250
E <sub>9</sub>	500	1000	1000
E <sub>10</sub>	500	1000	1000
Drug	Micromolar ( $\mu\text{gmL}^{-1}$ )		
NYSTATIN	100	100	100
GRESEOFULVIN	500	100	100

## CONCLUSIONS

The present work was synthesis of N-(4-(5-(substitutedeneamino)-1,3,4-oxadiazole-2-yl) phenyl) substitutedimine. Synthesised molecule identified by spectral data IR, <sup>1</sup>H NMR, <sup>13</sup>C NMR and Mass. From synthesised molecules performed antifungal, antibacterial and antitubercular biological studies. Some compounds give excellent biological activities and some gives sufficient activities.

## ACKNOWLEDGEMENTS

I am extremely thankful to Dr. D. Rajani, Microcare Laboratory, Surat, for screened biological activity and S. A. I. F. Chandigarh for <sup>1</sup>H NMR and <sup>13</sup>C NMR spectral analysis. I am also thanks to Dhanvantry, Kim for IR analysis and SynZeal, Ahamedabad for MASS Spectral data.

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#### AUTHOR ADDRESS

1. **Navin B. Patel**

Department of Chemistry,  
Veer Narmad South Gujarat University,  
Surat-395007, India  
E-mail: drnavinbpatel@gmail.com